



On the energy of curved carbon surfaces

A multi-scale Density Functional Theory and Monte Carlo study

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- ▶ graphene, (single and multiwall) nanotubes, fullerenes

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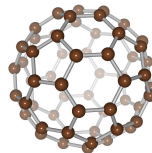
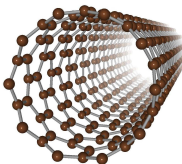
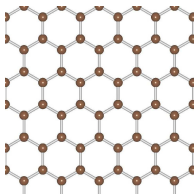
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- ▶ graphene, (single and multiwall) nanotubes, fullerenes
- ▶ superior mechanical and electronic properties

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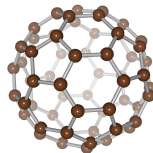
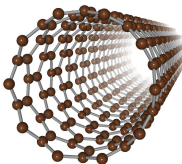
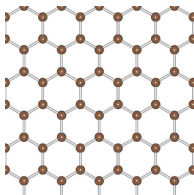
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- ▶ graphene, (single and multiwall) nanotubes, fullerenes
- ▶ superior mechanical and electronic properties
- ▶ fullerene C₆₀ first discovered in 1985 by Kroto *et al.*

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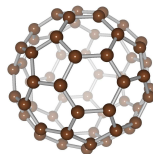
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- ▶ fullerene C_{60} first discovered in 1985 by Kroto *et al.*
- ▶ carbon aggregates tend to be more stable in a form of onion-like structures

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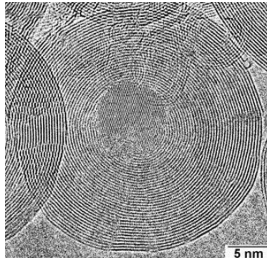
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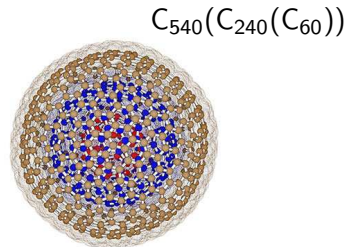
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Banhart *et al.*,
Phys. Chem. Lett. 269 (1997), 349.



<http://www.3dchem.com/>



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- ▶ superior mechanical and electronic properties
- ▶ fullerene C_{60} first discovered in 1985 by Kroto *et al.*
- ▶ carbon aggregates tend to be more stable in a form of onion-like structures
- ▶ possible use: supercapacitators, hydrogen storage

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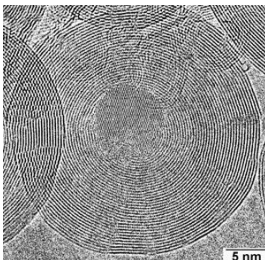
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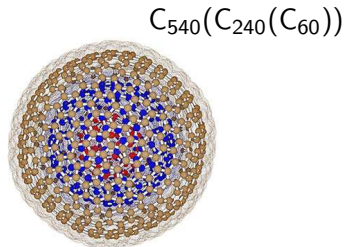
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- ▶ experiments suggest that the maximum number of layers is limited to $\approx 40 - 60$.

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- ▶ experiments suggest that the maximum number of layers is limited to $\approx 40 - 60$.

? Why?

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- ▶ experiments suggest that the maximum number of layers is limited to $\approx 40 - 60$.

? Why?

? May this be explained by mechanical instabilities occurring after a certain growth limit?

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- ▶ experiments suggest that the maximum number of layers is limited to $\approx 40 - 60$.
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- ▶ a continuum mechanics model of mechanical instability proposed by Bitsche *et al.* (submitted)



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- ▶ experiments suggest that the maximum number of layers is limited to $\approx 40 - 60$.
 - ? Why?
 - ? May this be explained by mechanical instabilities occurring after a certain growth limit?
- ▶ a continuum mechanics model of mechanical instability proposed by Bitsche *et al.* (submitted)
 - ▶ input: surface stress \approx surface energy
- ▶ the (excess) surface energy of a fullerene (or SWNT) is an increase of its total energy with respect to (planar) graphene



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- ▶ Vienna Ab-initio Simulation Package with GGA-PAW and LDA-US pseudopotentials



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- ▶ Vienna Ab-initio Simulation Package with GGA-PAW and LDA-US pseudopotentials



- ▶ only pentagons and hexagons \rightsquigarrow **exactly 12 pentagons, variable number of hexagons** (Euler's formula)

$$(\# \text{ of corners}) - (\# \text{ of edges}) + (\# \text{ of facets}) = 2$$

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- ▶ focus on fullerenes with **icosahedral symmetry** (with the exception of C_{70})

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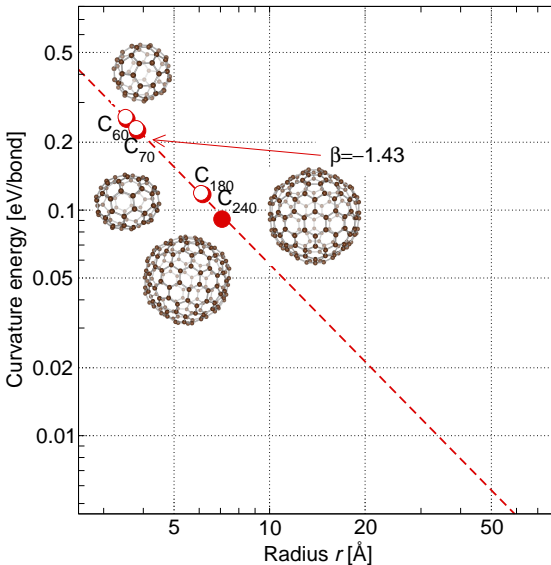
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radius:

$$r = \langle |r_i - \mathbf{R}_0| \rangle$$

power-law fit:

$$E = E_0 \times R^\beta$$

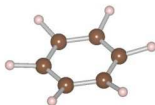
C_{240} :

12 days on

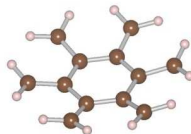
16 processors



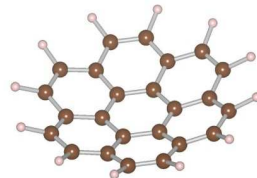
(a) one-ring shell



(b) two-ring shell



(c) three-ring shell



- ▶ parts of fullerene surfaces with dangling bonds
“saturated” with H atoms

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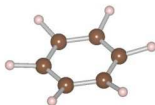
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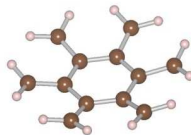
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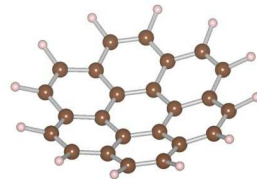
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- ▶ parts of fullerene surfaces with dangling bonds “saturated” with H atoms
- ▶ C–H bond length optimised to fit best the graphene energy (per bond)

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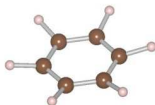
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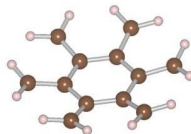
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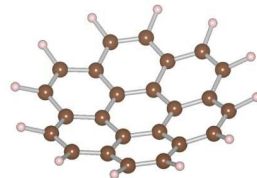
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+/- correspond to spherical structures

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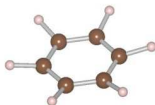
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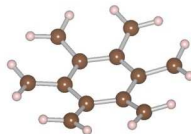
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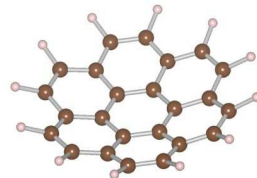
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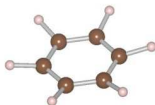
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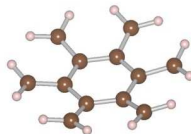
- ▶ parts of fullerene surfaces with dangling bonds “saturated” with H atoms
 - ▶ C–H bond length optimised to fit best the graphene energy (per bond)
- +/- correspond to spherical structures
- contain no pentagons



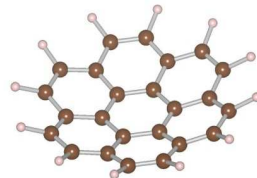
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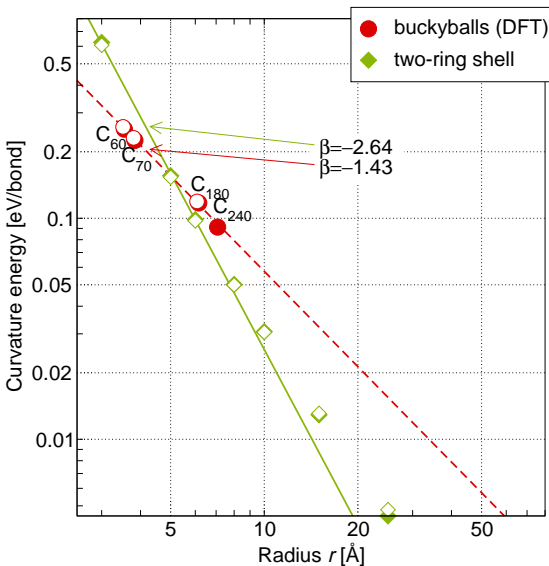
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- ▶ parts of fullerene surfaces with dangling bonds “saturated” with H atoms
 - ▶ C–H bond length optimised to fit best the graphene energy (per bond)
- +/- correspond to spherical structures
- contain no pentagons
 - + can easily access any desired radius

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radius:

$$r = \langle |r_i - \mathbf{R}_0| \rangle$$

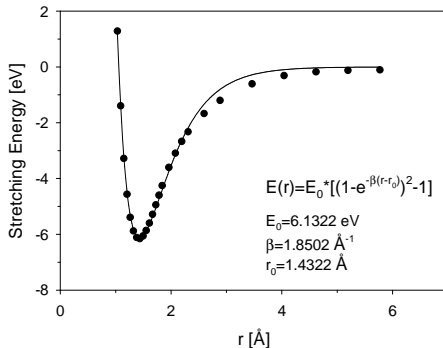
power-law fit:

$$E = E_0 \times R^\beta$$

construction of classical inter-atomic potentials “coherent”
with the present DFT calculations

- ▶ stretching: Morse potential

$$E^S(r_{ij}) = E_0 \left[(1 - \exp(\beta(r_{ij} - r_0)))^2 - 1 \right]$$



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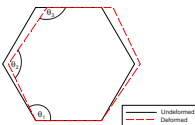
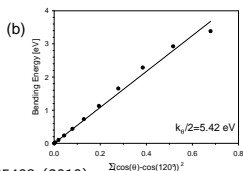
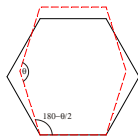
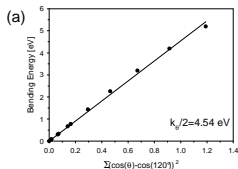
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construction of classical inter-atomic potentials “coherent”
with the present DFT calculations

- ▶ stretching: Morse potential
- ▶ bending: harmonic potential

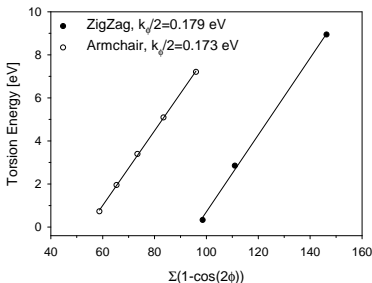
$$E^B(\theta_{ijk}) = \frac{1}{2}k_\theta (\cos \theta_{ijk} - \cos \theta_0)^2$$



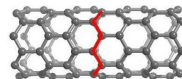
construction of classical inter-atomic potentials “coherent”
with the present DFT calculations

- ▶ stretching: Morse potential
- ▶ bending: harmonic potential
- ▶ torsion

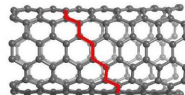
$$E^T(\phi_{ijkl}) = \frac{1}{2}k_\phi(1 - \cos 2\phi_{ijkl})$$



Zig Zag



Armchair





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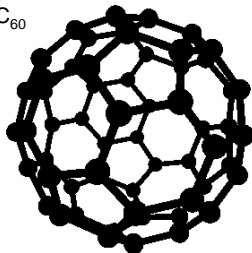
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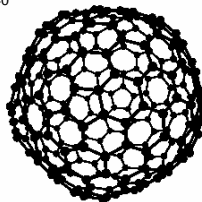
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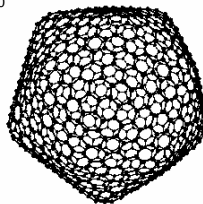
C_{60}



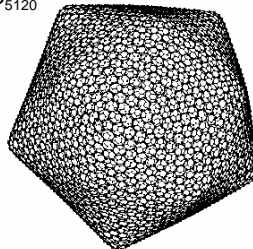
C_{240}



C_{980}



C_{5120}



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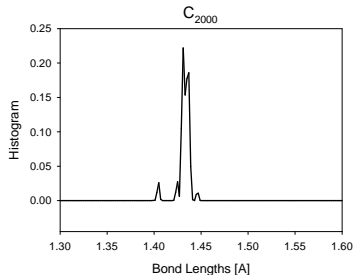
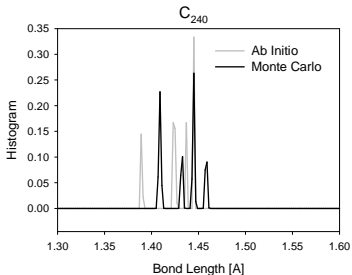
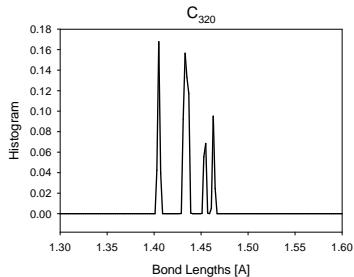
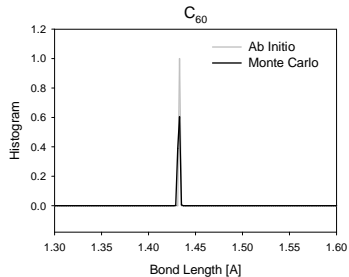
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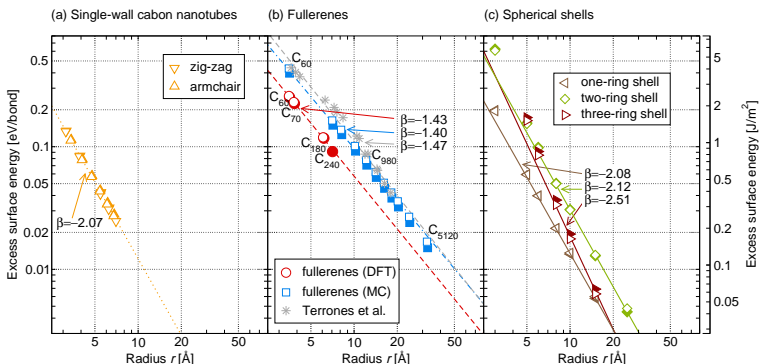
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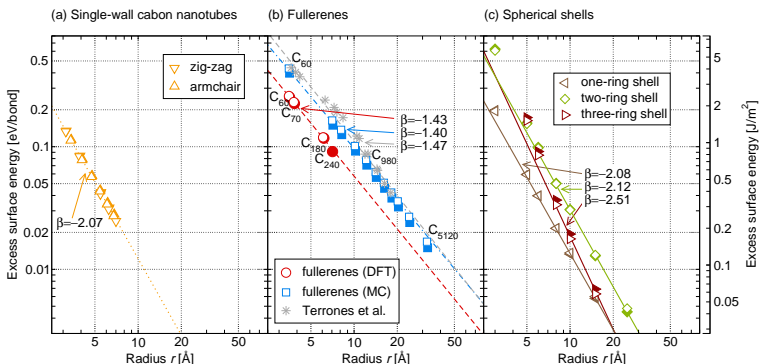
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- DFT and MC give the same exponent but different off-sets (power-law fit: $E = E_0 \times R^\beta$)



- ▶ DFT and MC give the same exponent but different off-sets (power-law fit: $E = E_0 \times R^\beta$)
- ▶ ring shell models fail to describe individual fullerenes but may be appropriate for (spherical) onion-like structures

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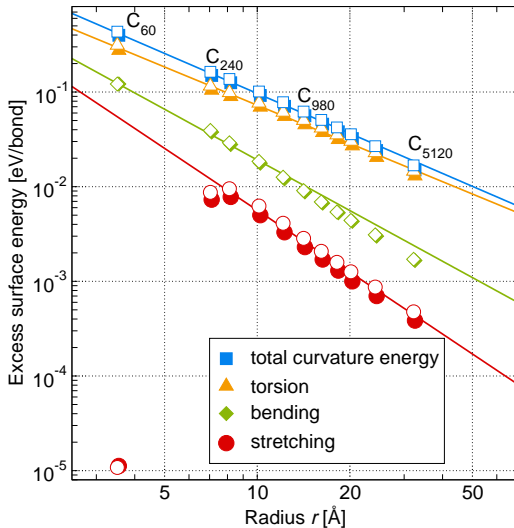
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- ▶ geometry and energy of carbon fullerenes were studied by a combination of the density functional theory and Monte Carlo methods

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- ▶ geometry and energy of carbon fullerenes were studied by a combination of the density functional theory and Monte Carlo methods
- ▶ DFT-compatible inter-atomic potentials for MC were constructed

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- ▶ curvature induced excess surface energy follows a power law as a function of the structure mean radius



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- ▶ DFT-compatible inter-atomic potentials for MC were constructed
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- ▶ power law exponent for the surface energy per bond:
 $\beta^{\text{DFT}} = -1.43$ and $\beta^{\text{MC}} = -1.40$

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- ▶ perfect spherical structures using approximative shell models yielded $\beta \approx -2.5$
- ▶ only stretching (two-particle) and bending (three-particle) interactions are not able to reproduce the correct fullerene geometries (i.e. bond-length and angle distributions leading to extended flattened areas), for which at least the four-particle interaction (i.e. torsion) has to be included.



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Methodology and Results

Fullerenes treated with DFT

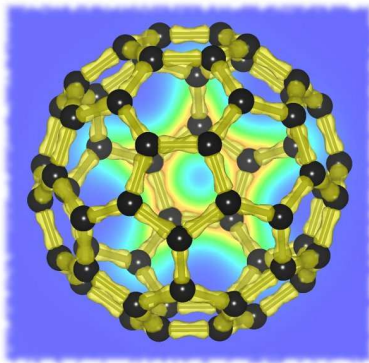
Ring shell structure

Monte Carlo simulations

Description of the fullerenes

Curvature energy

Conclusions



Thank you for your attention!

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